

IN THE SPECIFICATION:

Please amend the paragraph appearing at page 1, lines 3-9, so as to read as follows:

This is a continuation-in-part application of United States Patent Application Serial No. 10/384,020, filed on March 7, 2003 (now issued as United States Patent 6,693,160), which claims the benefit of United States Provisional Patent Application Serial No. 60/404,081, filed on August 16, 2002, and United States Provisional Patent Application Serial No. 60/434,892, filed on December 19, 2002.

Please amend the paragraph appearing at page 56, line 3 to line 17, so as to read as follows:

In this experiment 3-(2-pyrrolidinoethyl) and 4-(2-pyrrolidinoethyl) styrene were ~~synthesized~~ synthesized. In the procedure used 1030 grams of 80% divinylbenzene (824 grams of pure divinylbenzene 6.324 moles; the ratio of meta-DVB to para-DVB was normally 60:40) was added under nitrogen to a 5 liter flask equipped with a stirrer that contained 2 liters of dry hexane. To this homogenous solution was added 6.239 moles (450 g or 528 ml of dry pyrrolidine). This homogenous solution was cooled with wet/ice acetone to -5°C. At this temperature, 2.5% of the 6329 mmoles which is 155.9 mmoles of n-butyllithium was added all at once. The reaction temperatures rose to +55°C. The reaction was allowed to cool to +5°C for one hour. After that the reaction was neutralized with distilled water and three samples were taken for gas chromatography (GC) analysis.

Please amend the paragraph appearing at page 58, line 21 to line 30, so as to read as follows:

The PMS-styrene-butadiene terpolymer produced was determined to have a glass transition temperature (T_g) at -34°C. The Mooney viscosity (ML-4) at 100°C for this polymer was determined to be 37. The GPC data of this polymer was also determined to have a Mn of 147,100 and Mw of 180,600. The polydispersity (Mw/Mn) was 1.23. The polydispersity of this polymer is significantly higher than that of PES-styrene-butadiene terpolymer obtained in ~~Example 1~~ Example 34 (~~1.23~~ 1.23 vs. 1.05), indicating side reactions might occur when PMS was used as a co-monomer.